

PRESLHY Safety of LH2 Storage

Online workshop on Safe Storage of Compressed Gas Hydrogen in road transport applications and related infrastructure

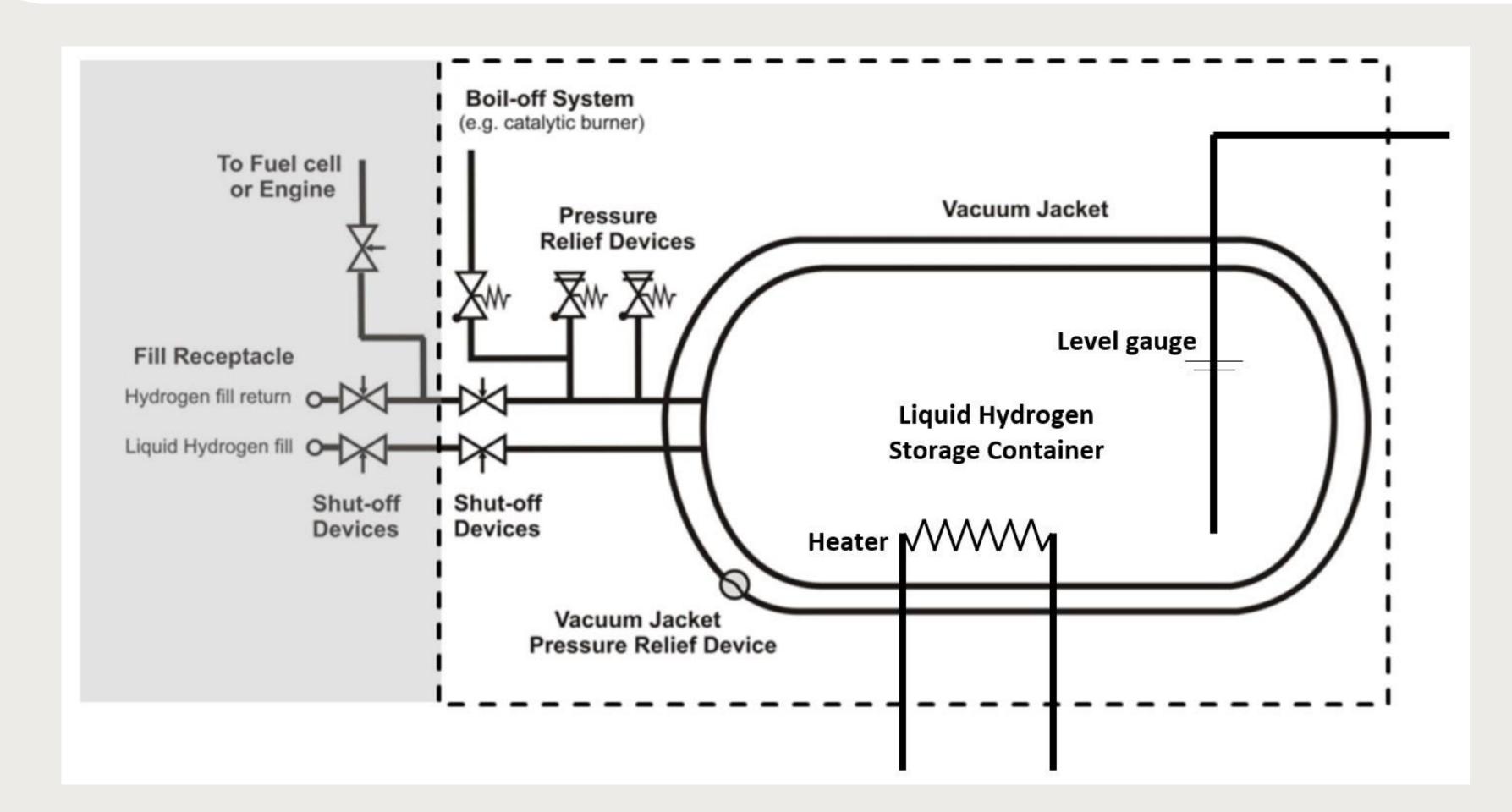


FUEL CELLS AND HYDROGEN JOINT UNDERTAKING

Thomas Jordan, KIT 18 November 2021



LH2 Storage







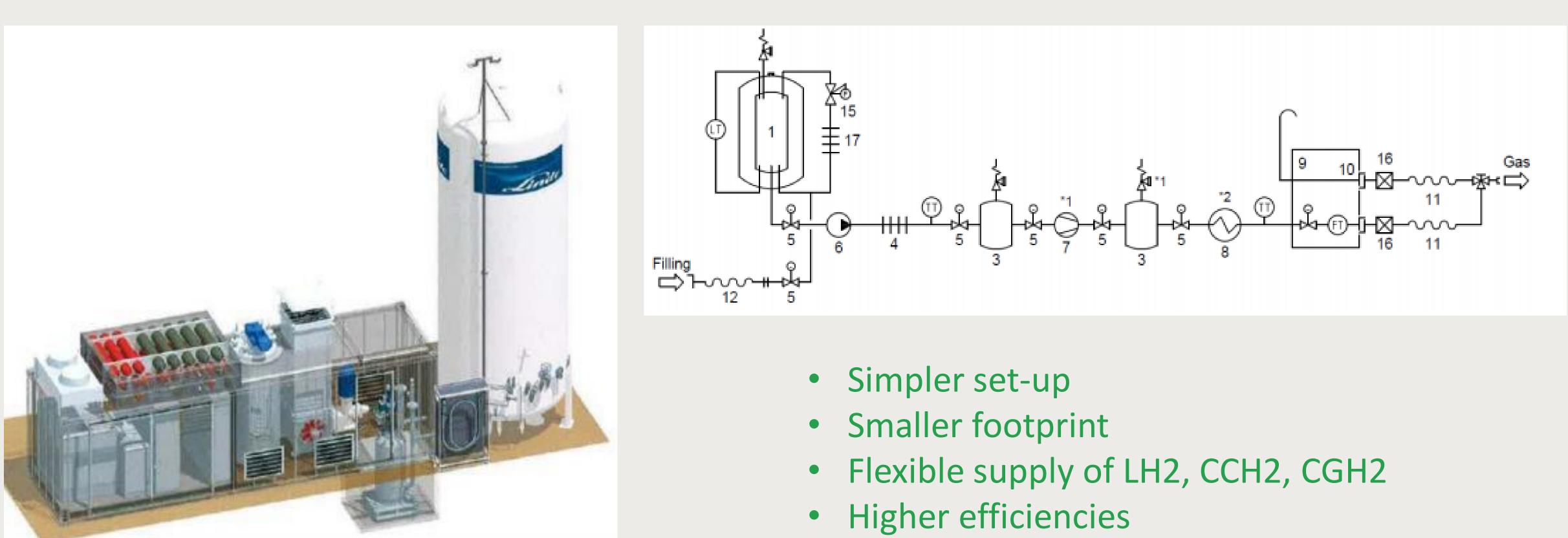


(Photo source HyCentA)





LH2 based fuelling station



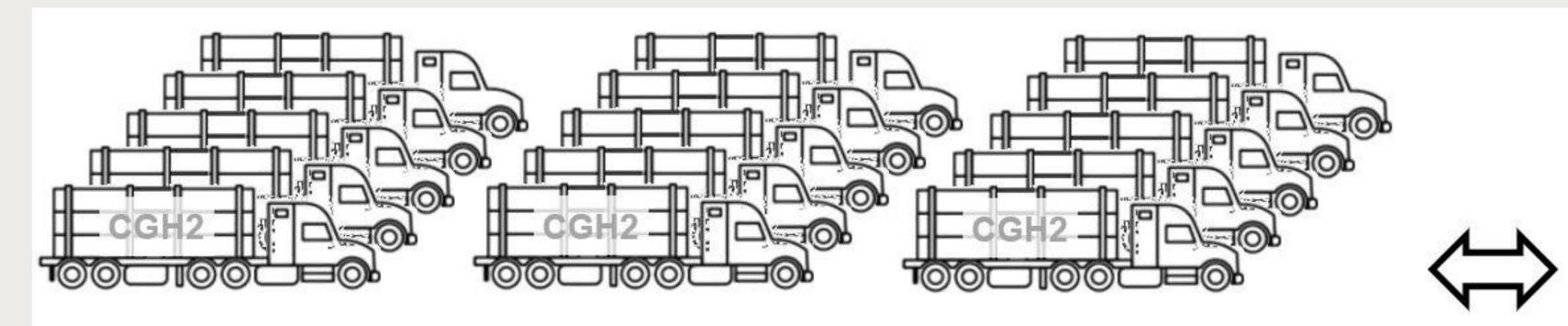
LH2 based HFS (courtesy of Linde)





- Potential multiple use of cooling capacity

Supply of a CGH2 vs LH2 based station



CGH2 20 MPa Type 1 250 kg capacity/trailer Trailer weight: 40 t ~1000 US\$/kg Invest

LH2:

- Reduces complexity
- Introduces cold gas venting (boil-off and flash gases)

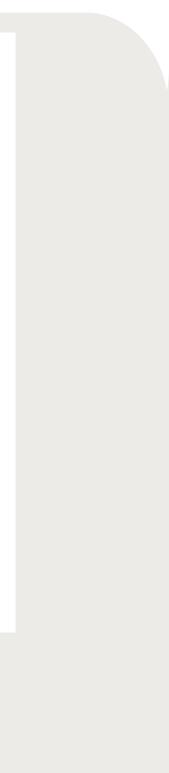




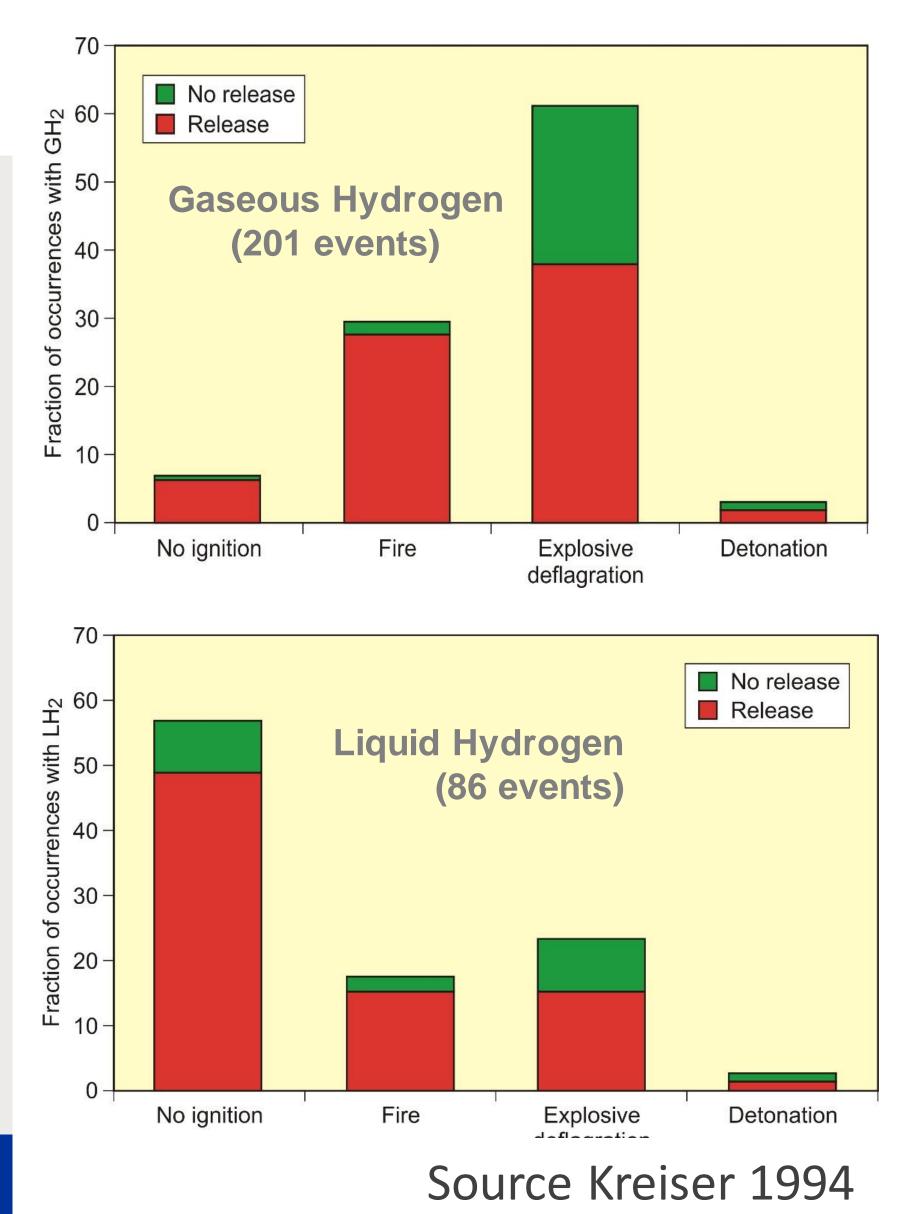


LH2 cryostat 4000 kg capacity/trailer Trailer weight: 20 t ~175 US\$/kg Invest

Reduces frequencies of transport, fillings with connections and de-connections Reduces weight and corresponding stresses on components and infrastructure



Accident statistics









Events around hydrogen road transportation

Design/Construction failure/inadequate Hazard Assessment	0 (0%)
Equipment failure	6 (33%)
Incorrect operation / procedural deficiency/poor maintenance	8 (44%)
Impact or Road Traffic Accidents RTA	3 (17%)
Contamination	0 (0%)
Natural causes/Terrorism	1 (5%)
Escalation	0 (0%)

Events around hydrogen liquefaction/storage

Liquefier/Purifier	2 (5%)
Vent system and pipework	11 (28%)
Storage vessels including fittings, valves and reliefs	14 (36%)
Valves/Components/Fittings	6 (15%)
Pumps/Compressors/Vaporizers	6 (15%)
Transfer lines/ pipelines	5 (13%)

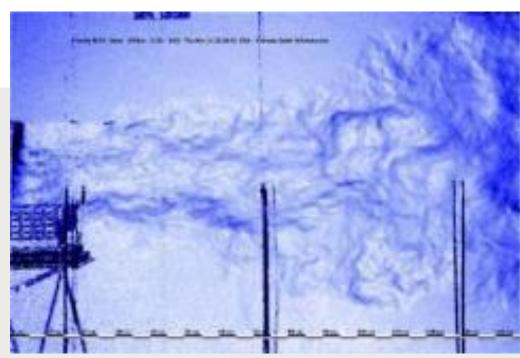
Closed Knowledge Gaps Release

- I D model for multi-phase release including non-equilibrium processes
- Discharge coefficients for circular nozzles D=0.5-4 mm 5 - 200 bar; 20 - 300K (KIT/PS E3.1 DISCHA tests) see https://doi.org/10.5445/IR/1000096833
- No rainout for large scale above ground horizontal releases (HSE E3.5: rainout tests)
- Correlation of T and concentration of mixtures of H2 with cryogenic origin and air



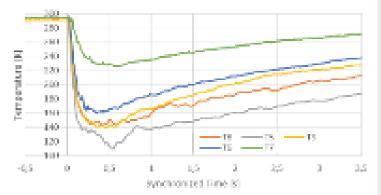


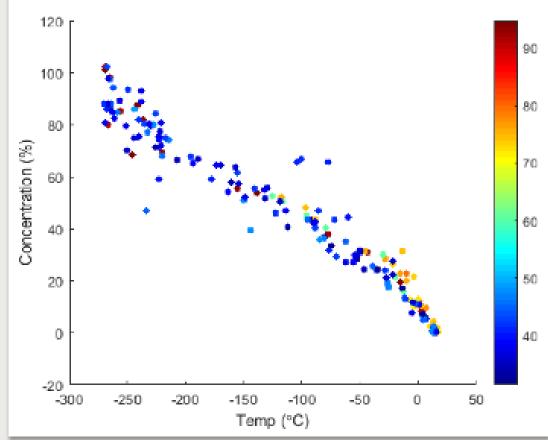




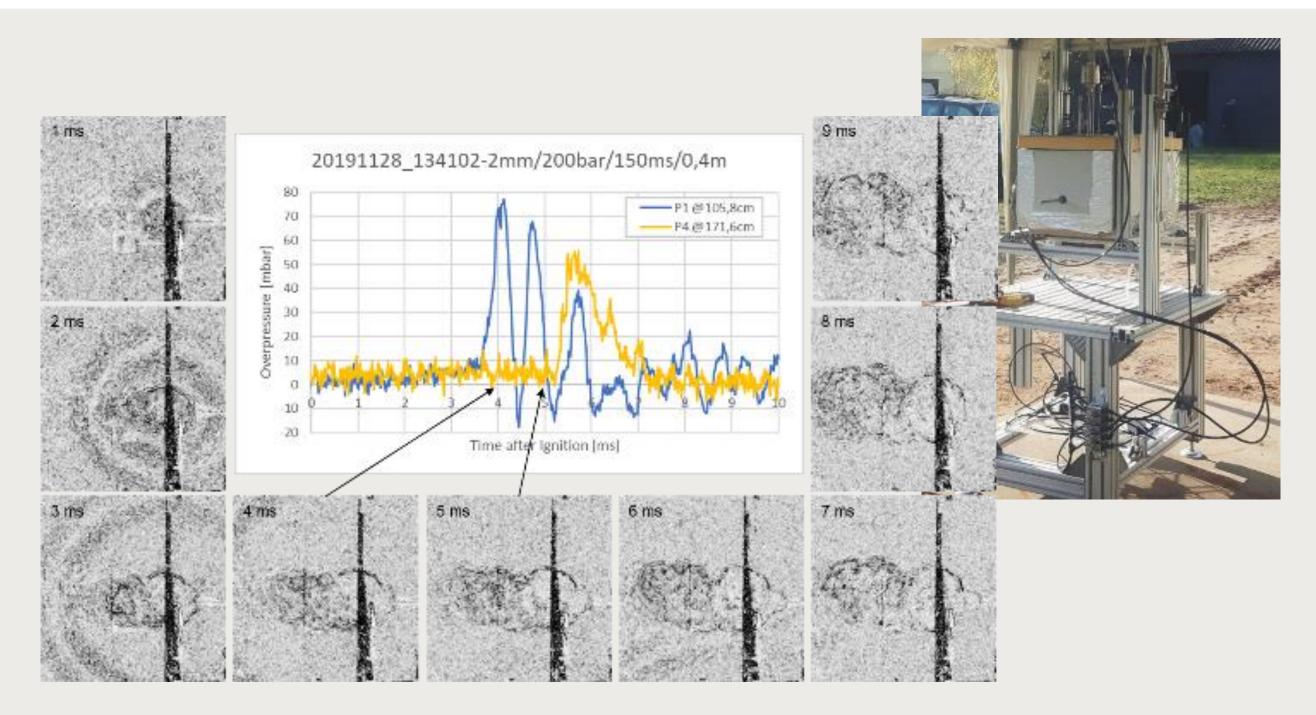
20190528_104204 (200 bar 4 mm 80 K) - p_v 120 -100Synchronized time [:

20190528 104204 (200 bar 4 mm 80 K) - T_w





Closed Knowledge Gaps Transient Combustion Effects



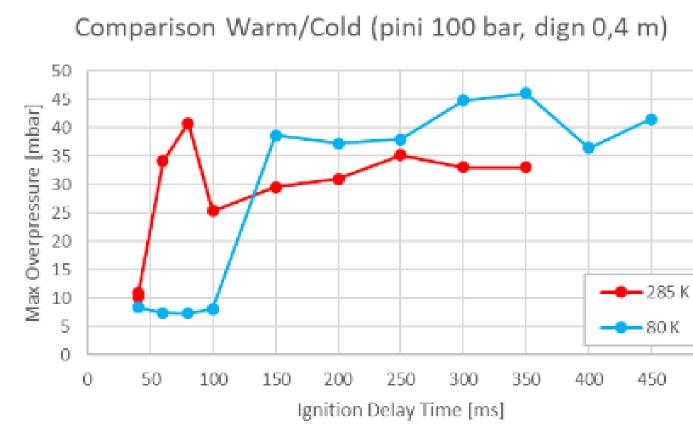
- Better understanding of transient jets and combustion processes
- Inventory based map of worst effects (pressure & thermal)
 - to be extrapolated to large inventories for RCS

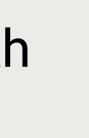


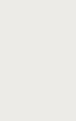


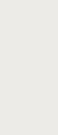
> 100 Ignited jet tests combined with discharge experiments E5.1 $T = 80K \dots 300K$ p = 5 .. 200bar $D_{nozzle} = 0.5 ... 4 mm$

Iterative procedure for identifying most critical ignition time and location

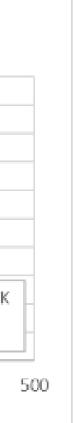












Closed Knowledge Gaps Combustion in confined/congested domains

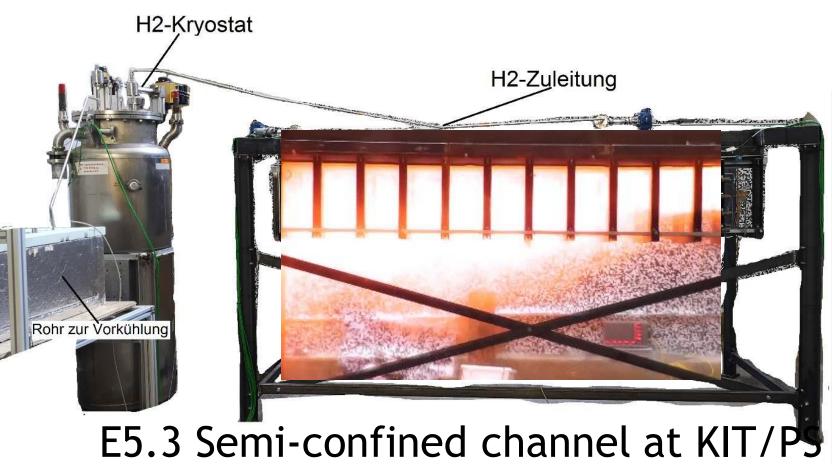
Stronger pressure loads for cold tests in comparison with warm tests with the same volume, hydrogen concentration and blockage ratio





E5.5 Test set-up at HSE, Buxton





Increase in critical and effective expansion ratios determine flame acceleration in cryogenic mixtures

Reduced run-up distance for detonation transition DDT in cryogenic mixtures (\leftarrow density effects)

Influence of blockage ratio on DDT less pronounced

Effects in free unconfined domains to be investigated

Closed Knowledge Gaps Multi-phase accumulations with explosion potential

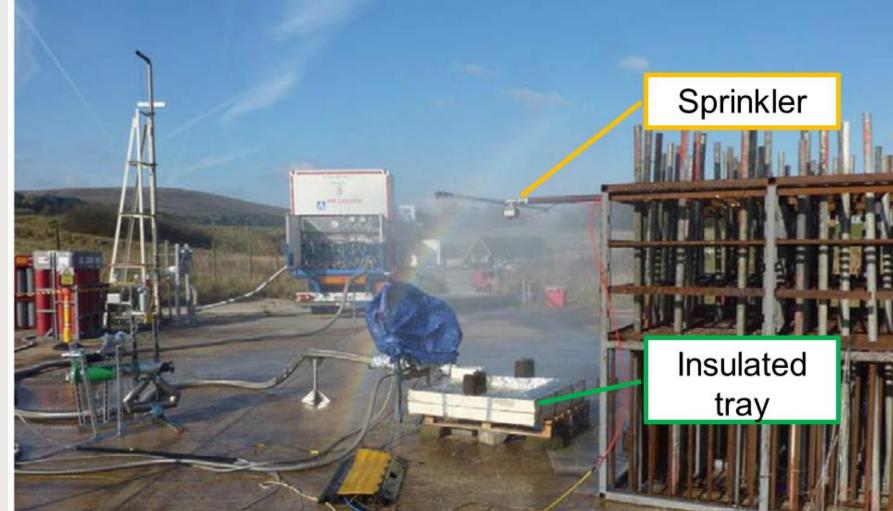
Repeated spills in gravel bed might generate highly reactive condensed phase mixtures not on other substrates (E4.4 Ignition above pool)

Water sprays on LH2 and LH2 spills on small water pools non critical (E4.4 and E4.X)



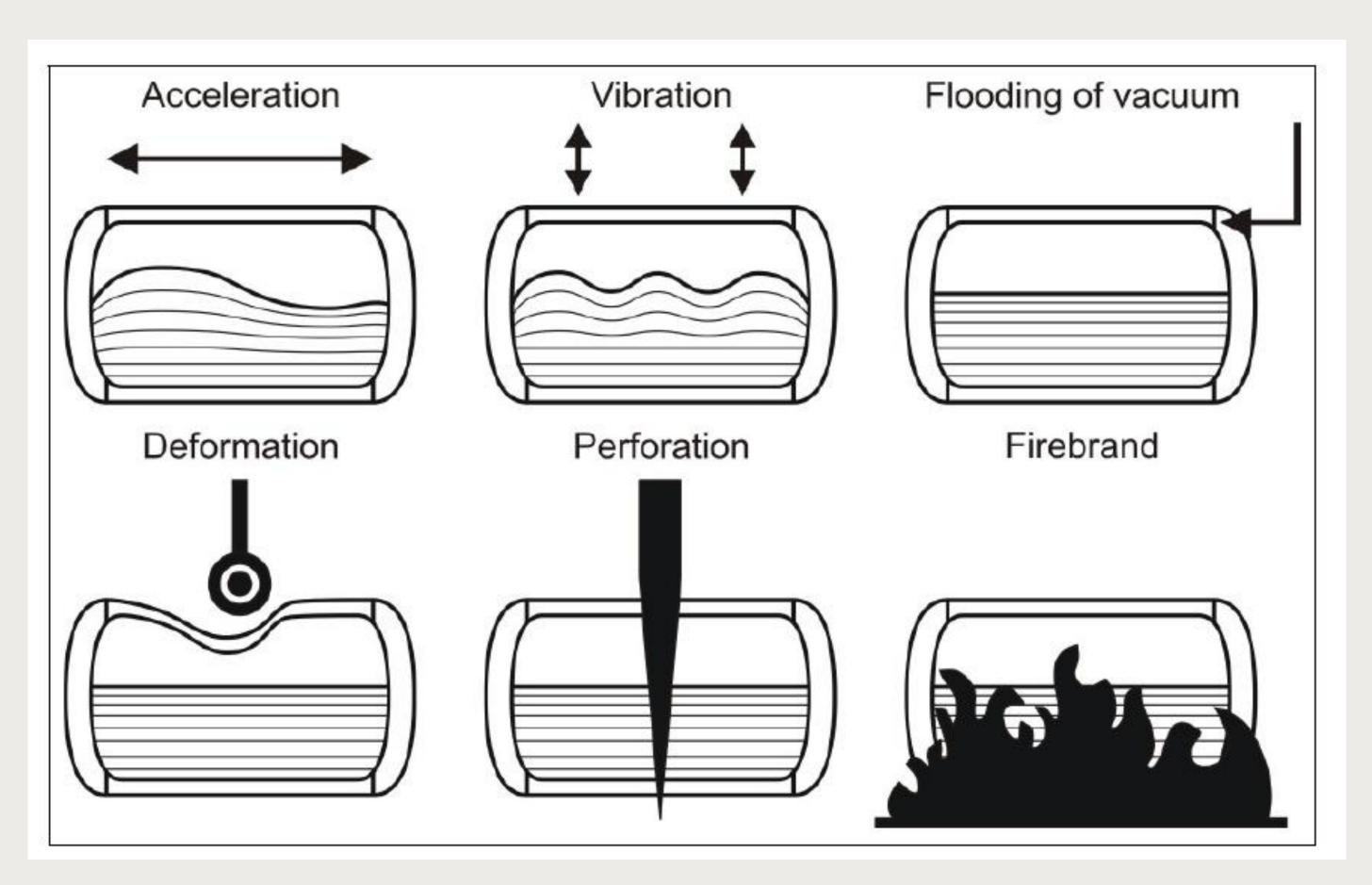






Possible Loads on LH2 Tanks

Actually developed for vehicle tanks

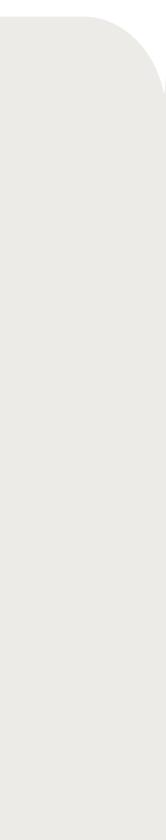






Source: Pehr 1995







Open Knowledge Gaps BLEVE

- Requires failure of heat insulation and the pressure envelop
- Release of mechanical, chemical and thermal energy (latent heat)





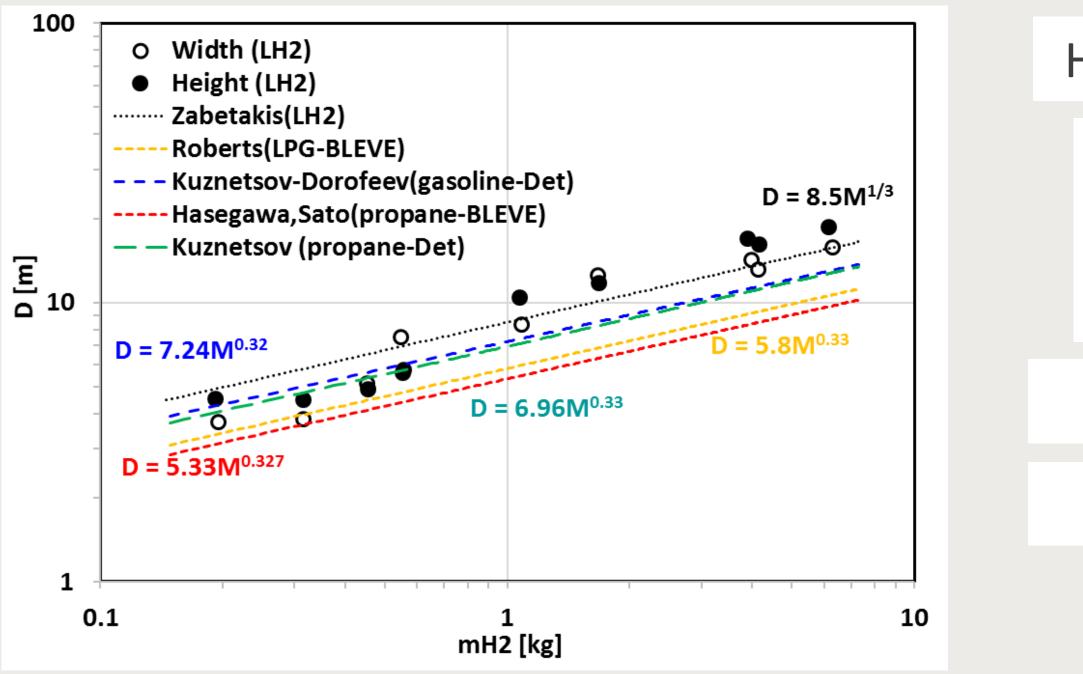






1998 BLEVE at Buchon South Korea LPG Gas station

BLEVE Scale correlations



 Safety distance is proportional to the mass of hydrogen







- Hydrocarbons
- $D=5.33 \cdot M^{0.327}$
- $t_d = 0.45 \cdot M_f^{1/3}$.
- $E = 8.085 \cdot M_{f.}$
- Hydrogen
- $D=8.5M_{f}^{1/3}$
- 10%H2-air

- 1.6 times larger LH2 fireball size than for hydrocarbons
- D=25.9m compared to calculated D=28 m for SH2IFT project (m=35.4 kg H2)

